STUDIES ON THE CONSTITUENTS OF HIBISCUS ROSA-SINENSIS

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A new flavone derivative, furanoflavone (1) was isolated along with ovalichalcone (2) from the leaves and stems of *Hibiscus rosa-sinensis* (Malvaceae). The structure of furanoflavone (1) was identified as 5-methoxy-3'-methyl-3',4'-methylenedioxyfurano[2", 3": 7,8]flavone by spectroscopic and chemical analysis.

Key words: Hibiscus rosa-sinensis, Isolation, Characterization, Chalcone, Flavone.

Introduction

The flavonoid constituents of the leaves and stems of Hibiscus rosa-sinensis continue to attract attention not only for their remarkable taste properties (Guadagni et al 1973 & 1976; Dubois et al 1982) but also for their therapeutic and pharmacological activities. They have been shown to possess biological activities such as anti-carcinogenic effect (Kato et al 1983; Nishino et al 1983; Harborne et al 1988; Venna et al 1988; Weitt et al 1990; Deschner et al 1991), anti-inflammatory properties (Busse et al 1984; Laudolfi et al 1984) and inhibitory activities against histamine release (Middleton et al 1984; Bronner et al 1985). The ethyl acetate extract of the leaves and stems of the plant yielded a novel furanoflavone (1) while the BuOH extract of the leaves and stems gave the ovalichalcone (2). The structures were identified on the basis of UV, IR, 1H-NMR, mass and gas liquid chromatographic data. Compound 1 identified as 5-methoxy-3-methyl-3',4'methylenedioxyfurano [2", 3": 7, 8] flavone is a novel natural product.

Experimental

Melting points were determined on an electrothermal melting point apparatus (Gallenkamp) which are uncorrected. IR spectra were recorded (KBr discs) on a Pye-Unicam SP3-300 IR spectrophotometer (v_{max} in cm⁻¹), ¹H-NMR spectra were recorded on a Perkin-Elmer R-32 (90 MHz) instrument in CDCl₃ with TMS as an internal standard (The chemical shifts are given in δ values). UV spectra were recorded on a Shimadzu UV -1201 spectrophotometer "Ultraspeck" in methanol. TLC were performed on silica gel plates. Mass spectra were recorded on VG 7070 E analytical mass spectrometer. A Varian gas chromatograph, Model 3300, equipped with Flame Ionization Detector (FID) was used for the GLC

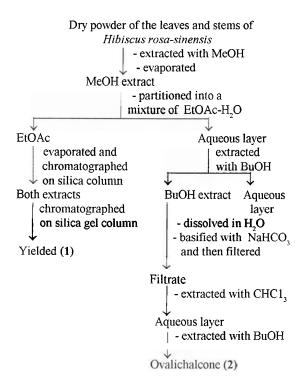
analysis. Singles were recorded on a Integrator, Varian model 4290. Hydrogen generator, model 7526, Canberra Industries Inc. was used for hydrogen production. A megabore column, SF 254 phase, 30 m long with 0.54 mm id., 1.2 μ film thickness was used for the chromatographic separation. Operation conditions:- Column temperature 90°C, detector temperature 300°C, injector temperature 250°C, carrier gas (N_2), flow rate 18 ± 1 ml min⁻¹.

Plant material: The leaves and stems of Hibiscus rosasinensis, Malvaceae were collected from a 10 meter high tree in the vicinity of the Bangladesh Atomic Energy Commission building in Dhaka (Bangladesh) in March 1998.

Isolation of Furanoflavone. The dried leaves and stems of Hibiscus rosa-sinensis (1.4kg), were powdered and separately extracted four times with MeOH under refluxing conditions for 18 hr. The extracts were concentrated and partitioned into a mixture of EtOAc and H,O. The water layer was washed with EtOAc and extracted with BuOH. In this experiment, EtOAc-extracts (5 g from the leaves and 4.2 g from the stems) and BuOH-extracts (3.5 g from the leaves and 2.9 g from the stems) were obtained. The EtOAc extracts from the leaves and stems were combined, due to the identical TLC behaviour. This extract was chromatographed on silica gel column and the following eluates were obtained: hexane (1.8 g), hexanebenzene (4:1) (1.6 g), hexane-benzene (1:1) (0.8g) and benzene (1.2 g), CH,Cl, (3.1 g) ether (1.2 g) and EtOAc (0.2 g). The benzene and CH,Cl, eluates were combined and chromatographed repeatedly on a silica gel column, and the eluates with benzene: EtOAc (10: 1) were crystallized from petroleum ether (b.p. 40-80°C) to obtain yellow prisms of compound 1, m.p. 161°C, yield (49 mg, yield 11.4%).

5-Methoxy-3-methyl-3', 4'-methylenedioxyfurano (2",3":7,8) flavone (1). Prism from petroleum ether 161°C; (M*, m/z 350); UV: 232, 250, 364 nm; IR: (KBr, disc): 2840,

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Scheme 1. Isolation of furanoflavone.

2210, 1642, 1605, 1595, 1475, 1363, 840, 740, 710 cm⁻¹; ¹H-NMR: δ 2.14 (s, 3H, -CH₃), 3.98 (s, 3H, -OCH₃), 6.01 (s, 2H, -OCH₂O-), 6.78 (s, 1H, H-6), 6.94 (s, 1H, H-2'), 7.12 (d, 1H, J=2.0 Hz, H-4"), 7.34 (d, 2H, J-9.0 Hz, H-5' and H-6'), 7.81 (d, 1H, J=2Hz, H-5"). (Found: C, 68.57; H, 4.00. C₂₀H₁₄O₆ requires: C, 68.66, H, 4.32%). It was identified as 5-methoxy-3-methyl-3',4'-methylenedioxyfurano (2",3": 7,8) flavone(1).

Results and Discussion

The dried leaves and stems of *Hibiscus rosa-sinensis* were separately extracted with boiling methanol and the extracts were treated as shown on Scheme-1. Furanoflavone (1) was obtained in the part soluble in ethyl acetate from the leaves

and stems as yellow prism. Its molecular formula was obtained by the elemental analysis and the mass spectrum (M^+ , m/z 350). The ethanolic solution of $\underline{\mathbf{1}}$ showed a negative ferric chloride test indicating the absence of free -OH group in the structure.

The IR spectrum of 1 showed the presence of carbonyl group at 1642 cm⁻¹ characteristic of flavone (Mukerjee et al 1969) carbonyl and the absorptions peaks at 1605 and 1595 cm⁻¹ indicated the presence of unsymmetric ethylenic double bond and the aromatic rings, respectively (Hossain et al 1998) The ¹H-NMR spectrum of the furanoflavone (1) showed three sharp singlets at δ 2.14 (3H), δ 3.98 (3H) and δ 6.02 (2H) indicating the presence of one methyl, one methoxy and one methylenedioxy groups. Two doublets at δ 7.12 and δ 7.81 indicated the presence of vinylic H-4" and H-5" protons on the furano ring (Hossain 1999). The remaining aromatic protons have usual chemical shift values. The known compound ovalichalcone (Islam et al 1993) was identified by comparison of its spectral properties with those reported in the literatures (Islam et al 1993). This compound was earlier isolated from Milletia ovalifolia. In the GLC chromatogram, the compound (2) had the retention time 9.56 min. which is same as the retention time of the standard ovalichalcone (2). The elemental analysis also showed complete agreement (within $\pm 0.4\%$).

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