

LOCAL DEVELOPMENT OF PEG ASSISTED DOUBLE ANTIBODY T₄-RADIONUCLIMUNOASSAY

Ehsan Mohammed Sojod

Malir Institute of Nuclear Medicine and Radiotherapy (MINAR), Malir, Pakistan

(Received 8 August 2000; accepted 25 May 2001)

In order to replace expensive ready-made commercial radioimmunoassay (RIA) kits with high quality local techniques a local Thionine (T₄) RIA was developed and evaluated. Four basic components of immunoassay system, antibodies, radiotracer, standards and precipitating solutions were locally prepared. Twelve rabbits were immunized to produce primary antibodies against T₄, using T₄ conjugated with bovine serum albumin (BSA) and emulsified with Freund's adjuvant as immunogen. Results showed that it out of 12 rabbits responded well to the immunogen and a maximum titer of 1:9700 was achieved with a working dilution from 1:1000 to 1:10000. Radiolabeling of Triiodothyronine (T₃) to produce radiotracer, ¹²⁵I-T₄ and ¹³¹I-T₃ by chloramine-T method showed that 31% of radioactivity was bound to T₄ and 50% to T₃ with specific activities of (11.1-14.8 MBq/ μ g) and (1.04-2.22 MBq/ μ g), respectively. Second antibody against T₄ was raised in two sheep. Results of optimization of second antibody showed a working dilution 1:60 for T₄ RIA. Precipitating solution to separate antigen antibody complex was prepared by mixing sheep anti-rabbit serum (SARS) with phosphate buffer containing 4% polyvinyl alcohol (PVA) at working dilution. PVA assisted double antibody T₄-RIA showed more than 75% error in the concentrations below 10 nmol/l with high precision at higher concentrations. The results of quality control samples showed that the values were within expected limits. The values correlated well with the commercial techniques (KEM, Coat. Coatf 10.99) and the values of patient samples determined in commercial as well as in house techniques did not differ significantly ($p>0.5$). The sensitivity of T₄ assay system was 2 fmol/l. It is, therefore, concluded that in house assay technique has acceptable quality and fits on all standards of good quality.

Key words: Local development, Radioimmunoassay, Separation.

Introduction

Immunoassays are most widely used analytical tools and have been successfully applied to an extensive range of substances, including both large and small molecules, cells, cellular components and viruses. Their applications in estimation of hormones, tumor markers and drugs in body fluids have greatly influenced the diagnosis and therapeutic strategy. Although the historical development has spanned over a century, the immunoassay techniques became very popular in last three decades. Initially immunoassays with radioactive and non-radioactive tracers were developed at almost the same time approximately 40 years ago (Berson and Yalow 1957; Berson and Yalow 1959; Blaustein et al 1968). The radioimmunoassay techniques became more popular because of good sensitivity, specificity and precision. In the late 80's non-isotopic techniques were also encouraged at many laboratories of the world. There were however sufficient reasons for retaining the use of isotopic techniques (Marks 1983; Turner 1988; Edwards 1994) which still compete well with the non-isotopic methods (Sojod 1999).

An immunoassay is based on interaction between analyte (labeled and unlabeled) and a specific antibody. The result-

ant antigen antibody complex can be separated from unreacted (or unbound analyte) by a number of separation systems. However, all of them have some disadvantages. For example systems based on differential migration are expensive and not very practicable, whereas those based on adsorption and flocculation precipitation are associated with significant misclassification errors (Obell 1980; Chard 1980) which results in inefficiency and lack of ruggedness. The most suitable technique has therefore to be selected after comparison with other techniques.

The labeled analyte is prepared by labeling of the analyte (which may be synthetic or extracted from human tissues) with an isotope. Initially these isotopes were preferred which are most common elements of biological fluids. For example ³H and ¹⁴C but mostly they required liquid scintillation counting which is inconvenient and expensive in comparison with the measurement of gamma radiation and needs scintillation well counters. The low energy of beta radiation of ³H also reduces the counting efficiency. The low decay rate of ³H and ¹⁴C with long half lives gives them low specific activity and correspondingly low count rates. This reduces their utility for high sensitivity assays. All other isotopes occurring in

biological materials also have suboptimal characteristics for use in radioligand assays. Either the half-life is very short, which makes their use impractical, or the type of radiation is less suitable e.g., ^{32}P with its high beta energy. Certain biological compounds contain other elements with radioisotopes that are more suitable for *in vitro* detection; examples are cyanocobalamin labeled with ^{57}Co and thyroxine hormones in which normal ^{131}I can be exchanged for ^{125}I or ^{131}I in more suitable because of low gamma energy (caused by electron capture; 35 keV) and long half-life (ca. 8 days). The isotope is usually supplied in the specific activities of 34-92.3 MBq/ μ mol (355-340 MBq/ μ g) and is attached to the analyte by oxidation of iodide to reactive atomic iodine which replaces the iodine atom of the molecule to be labeled (for example in iodination of T₄ and T₃). ^{125}I can also be used to indicate those compounds which contain ring structures of tyrosyl residues (similar to that found in T₄ and T₃). Here ^{125}I replaces one or two hydrogen atoms of ring structure; iodination of TSH with ^{125}I is an example. Almost all proteins contain tyrosyl residues and can therefore be labeled with ^{125}I . The antibodies (immunoglobulins) are also proteins and can be labeled with ^{125}I for the development of immunoradiometric assays (IRMA) (Jolani and Hunter 1973).

The binding reagent or antibody is a very important component of an assay system. The antibodies or immunoglobulins are produced by B-lymphocytes. After administration into an animal it is believed that immunogen binds to the receptors on the surface of lymphocytes. This binding induces the production of immunoglobulins with a surface of the variable portion that can bind this particular antigen. Then the lymphocytes carrying this information start multiplying, and all the new lymphocytes so produced carry with them the ability to produce antibodies against this particular immunogen. The basic structure of all immunoglobulin is similar with a pair of large polypeptide chains (heavy chains, Hc) that consist of 446 amino acids, each associated with a smaller chain (light chain, L) 214 amino acid long. There are five known classes of immunoglobulins IgG, IgM, IgA, IgD, and IgE. It is IgG which is important in immunoassays. The chains are joined together by disulfide bonds (Fig. 1).

Each immunoglobulin molecule has the ability to bind two antigen molecules, one at each of the two-binding sites, which

are located at the outer open end of each pair of H-L chains. Differences in the chemical composition at these two ends of the molecules (the so-called variable portions of the chains) are the structural basis for the differences in the binding characteristics of different antibodies.

The quality of an assay is highly dependent on the properties of the antibody used in an assay system and is assessed by sensitivity, specificity, precision and accuracy. The specificity is predominantly a property depending on the nature of antibody. The sensitivity (minimum quantity detectable) is limited by two factors, the avidity of the receptor (antibody) and the specific activity of the labeled analyte or radiotracer. The specificity (ability of the assay to measure one specific compound and no other) denotes the uniqueness of binding and the avidity (firmness of the binding or energy in the binding between the ligand and the receptor) denotes the strength of binding. To be able to measure the minute amounts of hormones and other biologically reactive compounds with sufficient sensitivity, avidity has to be very high. Optimal antibodies in this respect are sometimes very rare, so the avidity of antibodies is usually the limiting factor for the sensitivity of an RIA. Precision and accuracy are determined mainly by performance of the assays.

In Pakistan, the immunoassay services were introduced in late 70s and the work has been limited to the use of commercial radioimmunoassay kits. Constant dependence on expensive foreign technology is a great load on country's economy. About 20,000,000 rupees (~\$2000 US\$) are usually required in our institute for the purchase of RIA kits for thyroid related hormones. This work was an attempt to establish local techniques of reagent production to save the foreign exchange. Many other associated benefits like early release of results and increased scientific skills were also expected from this work.

The strategy and production schemes applied for preparation of in house T₄ assay reagents were similar to those applied in the development of T₄ assay and described elsewhere (Sajid 2001). The work was started with the production of antisera against T₄ and proceeded through preparation of assay buffer, radiotracer, standards and optimization of conditions for antibody dilution. Initially 15% PEG precipitation was tried which was then replaced with PEG assisted second antibody precipitation. Standard quality control procedures were applied to confirm the quality of locally developed techniques.

Materials and Methods

Purification of T₄-antiserum: Preparation of immunoglobulin (immunoglobulin): T₄ (Dyrenium): SIGMA Chemical Company, USA (iii) FBS (Human Serum Albumin): SIGMA Chemical Company (iv) Dimethylformamide (SIGMA Chem-



Fig. 1. Structure of immunoglobulin (antibody).

(al Company) (ii). CDI (Morpho-CDI; 1-Cyclohexyl-3-(2-morpholinoethyl)-1-carbodiimide metho-p-toluene sulphonic; Aldrich Chemical Company, USA) (v). Adjuvants: Two types of adjuvants were used. Freund's Complete Adjuvant was used at the time of primary injection. Freund's Incomplete Adjuvant, this was used in booster injections. The composition is given later on.

Preparation of T_1 -cyclohexylimmunogen. Small molecules (hapten) like T_1 and T_2 (Molecular weights 650 and 777 respectively) can not induce immunogenic response directly. The approximate limit for immunogenicity is molecular weight 1000 daltons or 10000 daltons. If a hapten is chemically coupled to a large molecule, the combination of the hapten and the large molecule (usually a large protein) acts as a complete antigen or immunogen. In very early studies by Sheehan (1956) and Good (1964) used water soluble carbodiimides as coupling agents for the production of antibodies against T_1 and T_2 . These molecules are first conjugated to human serum albumin (HSA) or bovine serum albumin (BSA). Carbodiimide is used to induce peptide bond between T_1 (the carbonyl group) and primary lysine residues (amino group) of HSA or BSA (Herbert 1978; Thorell and Larsson 1978). Figure 2 schematically illustrates the reaction. The T_1 -HSA conjugate was therefore prepared using method similar to that described by Herbert (1978).

20 mg of T_1 (Thyroxine; SIGMA Chemical Company, USA) and 50 mg of HSA were dissolved in 5 ml of dimethyl-

formamide and 25 ml of distilled water respectively. 30 mg of morpho-CDI was added to the albumin solution. T_1 solution was then mixed dropwise with the albumin solution with continuous stirring. pH of the mixture was continuously monitored by a pH meter and maintained at 5.5 using either 0.01 M NaOH or 0.01 M HCl. After 10 minutes of incubation at room temperature an additional 10 mg of morpho-CDI was added to the incubation mixture, which was kept at room temperature in the dark. The final volume of this mixture was 52 ml having a calculated T_1 level of 804/175.8 nmol T_1 /ml (625,000 ng/ml or 20mg T_1 /12 ml). 5 ml aliquot of this solution was drawn, diluted to 12.5 ml with phosphate buffer pH 7.5, 0.02M, (PBS/SA) providing an approximate concentration of 322 nmol T_1 /ml (T_1 /200 ng T_1 /ml) and saved for T_1 -RIA to be performed later. The above mixture was then dialysed against continuously running water in the dark.

Dialysis procedure: A 15 cm long dialysis tube (Dialysis tubing, SIGMA, benzalkonium, avg. flat width 12 mm, inflated volume approx. 100 ml/ml, suitable for retaining compounds with molecular weight over 2000 daltons, human serum albumin has mol wt 680000) was soaked in distilled water for 1 h. One end of the tube was closed with a knot. The other end was closed by inserting a 10 mm diameter rubber stopper through which a cannula was placed. The tubing was tied to the stopper with a string. One end of this string was used to fasten the dialysis bag with a stand to keep it suspended in the beaker containing distilled water. A small weight was also suspended at the lower end of the bag.

The solution to be dialysed was injected via the cannula into the dialysis bag. The dialysis bag was immersed into 1000 ml of distilled water stirred continuously using a magnetic stirrer. The water was changed after every 6 hrs. The dialysis was carried for 72 hrs. The volume of the dialysed mixture was made upto 12 ml using distilled water. 5 µl of this mixture was also drawn, diluted to 12.5 ml with phosphate buffer and saved for RIA.

T_1 concentration of the samples drawn before and after dialysis was measured by RIA using a commercial kit (Amersham, UK). The final mixture was aliquoted in 0.5 ml fractions in sterile vials, lyophilized and stored at 4 °C until use.

Preparation of emulsion for injection. T_1 -HSA conjugate was emulsified in a normal saline and Freund's adjuvant. The adjuvant acts as a non-specific stimulator to the lymphoid system, causing a foreign body reaction with granuloma formation. It also prolongs the adsorption of the immunogen, thereby extending the phase during which the lymphocytes of the host are exposed to the immunogen. Two types of adjuvants were used for immunizations in sup-

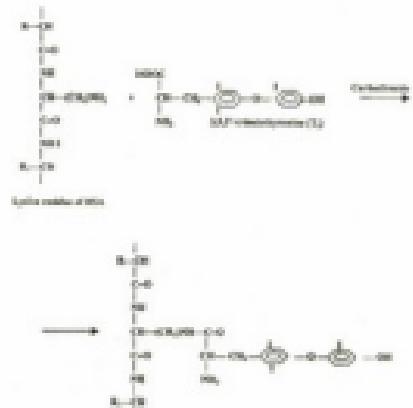


Fig. 2. Coupling of T_1 (a dipeptide) to a lysine residue of BSA by means of carbodiimide.

gested by Herbert (1978). The procedure of preparation was as under:

One ml of Freund's adjuvant (complete for primary injection) contains Artosol A, monamide monocarboxylate, Bayol F (paraffin oil), and heat killed and dried mycobacteria (*Mycobacterium tuberculosis*) and incomplete for booster injections (does not contain heat killed bacteria) and 1 ml of immunogen. Contents of one vial of T_1 -conjugate (prepared above) were reconstituted to 2.5 ml using isotonic saline to get immunogen at concentration, 0.3 mg/ml were drawn separately into two 10 CC syringes. Syringes were mounted against each other via the connecting piece (plastic tube). The contents were then pumped back and forth between the syringes until the mixture became thick. (It should be very thick, at least as thick as whipped cream or else the emulsion will not be stable). Since the plunger becomes difficult to press (into the syringe) when emulsion begins to thicken, the plunger was pressed against the surface of a laboratory bench for increased power. After about 5 minutes of pumping, a good emulsion was ready for use. To test the stability of the emulsion a drop of emulsion was allowed to float on lukewarm water and remain there for at least 5 minutes to be certain that oil and water phases do not separate. (Note: An unstable emulsion may give a poor immunogenic stimulus, since the water soluble immunogens are resolved rapidly. In case of immunogens with high biologic activity, this may even hurt the animals (Thorell and Larson 1978a, 1978b).

One milliliter (~150 µg) of this emulsion was injected to each rabbit. First responses are generally expected using low doses of immunogen. This also reduces the production of unwanted anti bodies (Munro et al 1983).

Animals, route and schedules of injection. Twelve rabbits were selected, weighing from 800g to 1.5 Kg. These were purchased from local market. Four rabbits were male and eight were females. These animals were kept and fed at institute's small animal house during the months from mid January to mid April and randomly labeled as Rabbit-1, Rabbit-2, — Rabbit-12. Wooden tags were hangs on their necks to maintain the identification. Usual green grass was provided as food. The floor of animal house was filled with natural soil to provide them natural environment. Proper arrangements were made for the ventilation and sunlight. Males were kept separate from females to prevent breeding.

Twenty multiple subcutaneous injections over the back of each rabbit were given. Immunogen emulsified in Freund's complete adjuvant was used for primary injection. In subsequent booster injections Freund's incomplete adjuvant was emulsified with the immunogen. The volume of each injection was around 40 µl. The back of the animal was shaved

prior to the injections (Munro and Wallace 1978; Butt et al 1983).

A 3ml blood sample was taken 30 days after the primary injection for testing of the antibody titre and at the same time a booster dose (immunogen emulsified in incomplete adjuvant) was given. Further booster doses were given at intervals of 30 days. The animals were finally bled 120 days after primary immunization. The blood was collected from ear vein by stabbing the ear with syrol. The serum was separated and preserved with 1% sodium azide.

Testing of T_1 antibodies in rabbits (dilution curve experiments). Antisera from 12 rabbits were collected and tested. For anti- T_1 titre by dilution curve experiments, Hare 15% coated polyethylene glycol (PEG) was used to precipitate antigen-antibody complex (Thorell and Larson 1978b; Cheung and Stanway 1976; Deshpande and Autioch 1971; Chard 1980). Procedure details are given below.

Reagents Prepared. (a) Assay buffer (phosphate buffer); pH 7.5, 0.05M, 17485A, sodium azide 0.05%; (b) Disodium hydrogen phosphate dihydrate ($Na_2HPO_4 \cdot 12H_2O$; MW=391.10), 17.85g, was dissolved in 900 ml distilled water with slight heating. The volume was made up to 1000 ml with distilled water. This was labeled as Solution-I. (c) Sodium di-hydrogen phosphate dihydrate ($NaH_2PO_4 \cdot 2H_2O$; MW=153), 1.53g was dissolved in approximately 200 ml distilled water and was made up to 250 ml. This was labeled as Solution-II.

The pH of solution-I was adjusted to 7.5 by addition of solution II. Test then sodium azide and BSA both in concentration of 0.1g/ml were added.

(d) 1% T_1 /ANS solution (~0.0001 cpm/10 µl). Diluted the concentrated solution of the tracer using assay buffer to get approximately 35000 counts per 50 ml volume. Added 2 mg per ml ANS (8-amino-1-naphthalene sulphonic acid) into this solution.

Thyroxine has very high affinity for plasma protein especially thyroxine binding globulin (TBG). About 80% of T_1 circulating in blood is bound to TBG which has a single high affinity binding site for T_1 . It is, therefore, necessary to block this binding site to get specific antigen antibody reaction. ANS is therefore used to inhibit this binding of thyroxine to plasma proteins in patient samples. Such binding interferes with antigen antibody complex. Thiomersal (mercuric nitrate), salicylate, natrichloride (tyrosine-β-halogenated tyrosine derivative), and phenoxine (diamine) have also been used for this purpose (Thorell and Larson 1978a, 1978c).

(e) 15% Polyethylene glycol, MW=6000 (PEG). 15g of PEG was dissolved in 100 ml assay buffer followed by the addition

of 0.1g Triton X100 (Edwards 1983). Polyethyleneglycol or PEG is a polymer that stays up in water and provides a well controlled antibody precipitation system. IgG is precipitated with 15% PEG.

Materials required: (i) Polystyrene assay tubes. (ii) Gamma Counter (Cap-Ria, USA, 16-channel). The counter was coupled with IBM compatible computer for data analysis.

Procedure for testing T_1 -antibody titre in rabbits using 15% PEG precipitation: Following dilutions of antisera were prepared using assay buffer: 1:200; 1:400; 1:800; 1:1600; 1:3200; 1:6400; 1:12800; 1:25600; 1:512000; 1:1024000; 1:2048000, and 1:4096000. Pipetted out 100 μ l of aliquots of these Ab-dilutions into appropriate tubes labelled in duplicate. Duplicate tubes were also prepared for total activity and non-specific binding (one separation of bound from free can be 100% complete and they will always be some residual free antigen concentration associated with the bound fraction). This is non-specific binding which needs to be estimated with considerable degree of accuracy as errors in it's determination have a disproportionate effect on low bound to free ratios in an assay. 100 μ l of assay buffer was added in these tubes. Then sequentially following additions were made in each tube: 50 μ l of hormone free serum per tube (this is to increase the yield of precipitates); 50 μ l of ^{125}I -ANS solution per tube.

After this all tubes were kept at room temperature (18–20 hrs) and then following day the mixtures were incubated at 4°C followed by the addition of 0.5 ml of 15% PEG into each tube. After thorough mixing with a vortex mixer, the mixtures were allowed to stand for 15 minutes at 4°C. Centrifuged the tubes for 30 minutes at 3000 rpm followed by decantation of liquid contents. The precipitates were counted for 90 seconds using a gamma counter and the bound counts were analyzed on the computer.

Preparation of labelled precursors for T_1 and T_2 RIA's. A variety of methods for iodination of T_1 and T_2 are available in the literature (Bar et al 1983). For this study we selected the method described by Edwards et al (1983) using Chloramine-T oxidation (Crossman et al 1963) because this method gives tracers of high specific activity. Moreover two isotopes (^{125}I -T₁ and ^{125}I -T₂) can be obtained in single iodination experiment followed by purification by gel filtration.

Principle and reaction mechanism: Radioactive iodine used for labelling is usually supplied in iodide form. The iodide is oxidized to a reactive cation, which attacks aromatic ring of tyrosine. The mechanism of reaction is explained below—

Radioactive iodine is incorporated into the aromatic ring of tyrosine in two step reaction. In the first step radioactive iodine (^{125}I) is oxidized to reactive cation ($\text{HO}^{125}\text{I}^+$) which then

attacks on the nucleophilic aromatic ring. The reaction is schematically explained in Fig 3.

The reaction conditions were maintained in such a way that one radioactive iodine atom is incorporated per molecule. The formation of molecules containing more than one radioactive atom may be unsuitable because of high radiation damage. The final mixture contains a number of species, which needs to be separated. The purification step is therefore essential and is explained in procedure given below.

Methodology of iodination by chloramine-T method

Reagents required: (i) Preparation of phosphate buffer (pH 7.3; 0.05M). Procedure of preparation of phosphate buffer was the same as described for preparation of assay buffer except that addition of sodium azide and BSA was excluded.

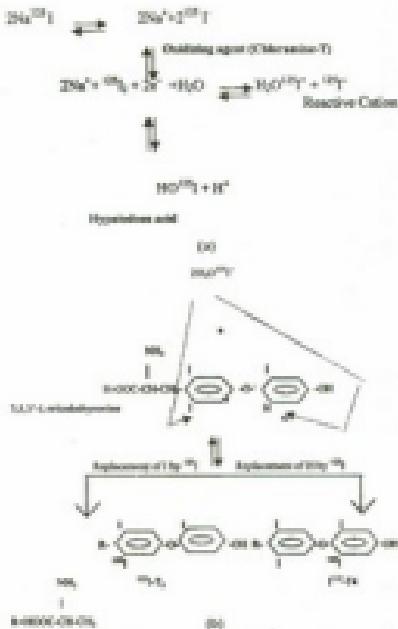


Fig 3. Mechanism of interaction of radioactive iodide with T_1 molecule (a) oxidation of radioactive iodide by chloramine-T method. (b) Electrophilic attack of reactive cation on aromatic rings.

(ii) T_1 Solution: Concentration = 1 $\mu\text{g}/5 \mu\text{l} (2 \text{mg}/\text{ml}), \text{pH} 8.0$. 2mg of T_1 was taken in a clean 10 ml graduated beaker or volumetric flask and 4 ml distilled water was added into it. Added 1M NaOH drop wise with stirring until a clear solution was obtained. Diluted the solution to 10 ml with distilled water. Adjusted pH to 8.0-8.2 using 1M NaOH/PO₄ solution taking care that no turbidity appears.

(iii) ^{131}I -NaI Solution, Concentration = 3mCi/50 μl . Provided by the manufacturer (Amersham International, UK). Centrifuged at 600 rpm for 5 minutes to remove any particles.

(iv) Chloramphenicol solution, Concentration = 20 $\mu\text{g}/5 \mu\text{l} (4 \text{mg}/\text{ml})$: 20mg of Chloramphenicol was added in 1 ml phosphate buffer pH 7.5. Stirred until dissolved. Made up the volume to 2ml.

(v). Na-Adrenalin/NaHBBG Solution, Concentration = 20 $\mu\text{g}/5 \mu\text{l}$ (4 mg/ml): 80mg of NaHBBG ($\text{Na}_2\text{S}_2\text{O}_5/\text{Na}_2\text{B}_4\text{O}_7$) was dissolved in 10ml phosphate buffer (pH 7.5) and made up the volume to 20 ml.

(vi). Potassium iodide Solution, Concentration = 10 $\mu\text{g}/\text{ml}$: Took 100 mg of KI in 1 ml phosphate buffer (pH 6.4, pH 7.2).

(vii). bicarbonate buffer: 50mM, pH 9. (0.05M of NaHCO₃/8.4); 4.2g of NaHCO₃ was dissolved in 900 ml of distilled water; pH of the solution was adjusted to 9 with 1M NaOH drop wise. The volume was made up to 1000 ml with distilled water.

(viii). Preparation of Sephadex G-25 (Sephadex): Columns: 8 g of Sephadex G-25-500 Superfine; Bead Size: 20-50 μm , (Pharmacia, USA) was taken and swollen up in 25 ml phosphate buffer over 18-24 hrs. Packed in 30 x 1 cm column size.

Equipment and disposables arranged were:-

- Single well gamma Counter
- Fraction Collector (Frac-500, Pharmacia, Sweden)
- Polystryrene tubes
- Poly ethylene gloves
- Micropipette (SOCTOREX, Germany)

Following reaction procedure was followed:-

Following reagents were added in following quantities into a clean micro test tube in rapid succession.

Phosphate buffer	10 μl
1 mCi ^{131}I -NaI/50 μl	10 μl and vortexed
1 μg $T_1/5 \mu\text{l}$	5 μl vortexed
20 μg Chl-TG/5 μl	5 μl vortex mixed for 20 sec
20 μg Na ₂ S ₂ O ₅ /5 μl	5 μl vortex
1000 μg KI/100 μl	100 μl vortexed

Applied the mixture to a Sephadex G-25 column (Column size: 30 x 1cm). Waited for 15-20 minutes to achieve adsorption equilibrium. Eluted the column with 50 mM NaHCO₃ (pH 9.0). Collected 2 ml fractions at setting 3 of peristaltic pump (flow rate 14 ml/hr). Counted the fractions using single well gamma counter. Plotted counts of above fractions (Table 1) versus tube number

Table 1
 T_1 levels in the dialysis mixture

T_1 concentration (nmol/l)	Recovery
Calculated value (before dialysis)	322
Measured value before dialysis	264 ± 18 (SD)
Measured value after dialysis	199.5 ± 07

(radiochromatogram) and identified three peaks (Fig-9). The first one was untracted ^{131}I -peak, the second being $^{131}\text{I}-T_1$ peak and the last one is $^{131}\text{I}-T_1$ peak. Pooled the desired peak fractions containing radioactive T_1 and ^{131}I . To each added 0.1 M NaHCO₃ until pH was 7.5. Eluted each pool to radioactive concentration of 3-10 pCi/ml, adding phosphate buffer, containing 1% albumin, 2% mannitol and 0.1% cysteine.

For preparation of working $^{131}\text{I}-T_1$ solution or tracer, diluted the concentrated solution of the tracer using assay buffer to get approximately 15000 counts per minute per 100 μl volume. Added 2 mg per ml of T_1 inhibitor: ANS into this solution.

Conditions for an ideal tracer: As already stated many methods of iodination are available but chloramine-T (Greenwood et al 1961) is the most widely used iodination procedure. The principles and requirements of the iodination reaction have been reviewed (Hunter 1977) and optimum conditions for some steps in the reaction have been investigated. An ideal tracer would be a homogeneous preparation of the antigen which has been labeled without loss of affinity for the antibody being used. Methods are available for checking the heterogeneity of the tracer, (Belton 1977, Belton and Hunter 1977) which results due to changes in the antigen. The changes in the antigen may result due to (i) the presence of the label (^{131}I), (ii) chemical changes resulting from exposure to reagents used in the labelling, such as the oxidant and reducing agent, (iii) handling and exposure to microgram quantities of highly purified protein and unprotected by carriers, (iv) radiation damage. Generally unexpected low binding in an assay indicates a damaged (heterogeneous) tracer.

Optimization of T_1 -antibody quantity for use in T_1 RIA. The most important step in designing a radioimmunoassay is to determine the antibody concentration, which gives an assay system of required sensitivity and precision in the desired clinical range. The experimental set up for this purpose would first require selection of optimum quantity of radiotracer, which is another parameter to adjust the sensitivity of an assay system.

Optimal concentration of radiotracer in our set up: Optimal amount of radiotracer would be that which could ideally saturate all the binding sites of the antibody at optimal dilution. However practically this is not possible. Efforts were therefore made to establish optimised conditions (Homer 1983). We have observed that simple arithmetic assessments are often very useful in selecting the optimum concentration of radiotracer. Following was our strategy to prepare workable solution of radiotracer for our assays.

The matter was started from expected or desired clinical which was decided on the basis of normal range reported in literature. In T₃, we were interested in setting up assay for levels within, above and below the normal range. Following calculations were therefore performed.

Expected normal range (Theorell and Larsen 1978c) = 45-120 ng/ml (58-154 nmol/l).

So the desired working range of the assay may be= (0-300 ng/ml (this should normally cover hypo, hyper and euthyroid subjects).

For a sample of volume, 50 µl, 10 ng T₃/ml will correspond to 0.5 ng T₃/50 µl (3000pg/50µl or 60000 pg/ml).

Thus we need to add 500 pg of ¹²⁵I-T₃ tracer into each tube in our assay system.

We therefore require a solution of concentration, 0.5 ng ¹²⁵I-T₃/50 µl.

For a specific activity (see iodination section) of 50 mCi/mg or 60 mCi/mg, this will correspond to 25 nCi ¹²⁵I-T₃/50 µl. The expected count rate can therefore be calculated as under:

$$25 \text{ nCi} = 3.7 \times 10 \times 25 \text{ dpm} = 55000 \text{ dpm}$$

dpm = dpm's efficiency

For a gamma counter of 70% efficiency,

$$55000 \text{ dpm} \times 0.70 = 38500 \text{ dpm}$$

Thus for our assay system we are interested in, we need to prepare a tracer of concentration, approximately 30000-40000 cpm/ 50 µl to get precise results in the region of desired working limit. This concentration, however, is an estimate and the actual concentration may be different because as already described there are many factors which lead to the reduced binding. These are related to the heterogeneity of the tracer. Rate of radioactive decay must also be taken into account.

Optimal concentration of antibody: Ideally an optimum concentration would be that which is completely saturated by the amount of tracer we have selected and prepared. However, in practical, this can not be achieved because the antigen-antibody reaction is affected by a number of factors (Theorell and Larsen 1978c). For example in many antigen-antibody systems primary reaction is followed by secondary

reactions. The primary antigen-antibody complexes join together and form aggregates (and these aggregates having large lattices reduce their solubility in water which cause them to precipitate). Other secondary effects of antigen antibody reaction are agglutinations and cell toxic reactions. In RIA it is assumed that only the primary reversible antigen antibody reaction occurs (except in separation with double antibody) and follows the law of mass action (Here an equilibrium between concentrations of the reactants (antigen and antibody concentrations) and the product (the antigen-antibody complex) is achieved. The optimal concentration of antibody would therefore be that which leads to the equilibrium and gives minimum error in the results. It can also be determined from dilution curve experiment. The procedural details are given below.

Dilution curve experiment: Briefly, following dilutions of antiseraum (rabbit-I) were prepared in assay buffer: 1:20; 1:100; 1:200; 1:400; 1:800; 1:1600; 1:3200; 1:6400; 1:12800 and aliquoted in excess of tubes for determination D and D₀. 100 µl of Ab-dilutions were aliquoted into appropriate tubes except NSB tubes where 100 µl of assay buffer was added. Other reagents were 50 µl of ¹²⁵I-T₃/NSB (x/50000 cpm) into each tube, 50 µl of hormone free serum into each tube of set-1, 50 µl of T₃-standard of conc: 20 nmol/l into each tube of set-2. The mixtures were vortexed and incubated at room temperature overnight. 0.5 ml of 15% PEG was added at the end of incubation. The tubes were finally centrifuged for 30 minutes at 3000 rpm, decanted and counted the precipitate for 90 seconds.

Production of sheep anti-rabbit serum-SABS (2nd antibody) for use in PEG assisted double antibody precipitation: Principle of PEG assisted double antibody precipitation: The non-specific binding is high when we use 10% PEG to separate antigen-antibody complex which is generally associated with significant misclassification errors, variable behaviour of damaged tracer, serum effects and interference in primary antigen-antibody reaction (Obell 1980; Chard 1980). This much non-specific binding could affect the working range of the assay. The precipitation with second antibody could overcome these difficulties. This increases specificity of separation of antibody, and hence the bound fraction, from the free fraction. At the end of the incubation, an anti-IgG antibody is added that is specific to the species in which the primary antibody was raised. The divalent nature of antibodies is responsible for the formation of a lattice consisting of cross-linked antibodies and analyte. More than one molecule of antibody may be bound to each molecule of analyte, at different epitopes, each molecule of primary antibody may bind to two molecules of analyte, and each molecule of second antibody can bind to two molecules of primary antibody. The lattice of cross linked molecules

precipitates and separation is achieved centrifugation and decantation. PEG assisted second antibody precipitation combines the benefits of second antibody and PEG methods: specificity and speed. The PEG which is used at lower concentration of 4%, hinders the precipitation of the cross-linking matrix (Wild and Davies 1994).

Second antibody is produced by immunizing a different animal species with immunoglobulin G (IgG) purified from the species in which the first or primary antibody was raised, giving an antiserum capable of precipitating the first antibody. In our case we used primary antiseraus antibodies. We therefore immunized sheep with rabbit IgG.

Preparation of immunosorber Reagents Prepared preserved immunogen (Rabbit IgG; Sigma, USA). IgG fractions of non-immune animal sera were prepared by the manufacturer by caprylic acid fractionation as described by Seibelsch and Andrian (1969) followed by DEAE ion-exchange chromatography. Immunogens thus prepared were assessed for homogeneity by immunoelectrophoresis as described by Erlanger et al (1959).

Adjuvants: (i) *Freund's complete adjuvant*: This was used at the time of primary injection. Standard Scottish antibody production (SAPD) protocol (Chapman et al 1983) was used i.e., 0.1 mg IgG in 2.5 ml phosphate buffer was emulsified in Freund's complete adjuvant and administered by multiple intramuscular injections. The total volume of emulsion was 5 ml. It was injected at 10 sites.

(ii) *Freund's incomplete adjuvant*: In booster injections same procedure was used but Freund's complete adjuvant was replaced with Freund's incomplete adjuvant. Booster injections were given at 1 month interval. The injection sites were near the primary injection sites.

Procedure of raising the antibodies: Rabbit IgG was dissolved in phosphate buffer to make a concentration of 0.25 mg/2.5 ml. One ml of this solution was mixed with Freund's adjuvant using the procedure described for anti-T₄ (and anti-T₃) antibody production.

Animals, route and schedules of injection: The choice of animals for production of second precipitating antisera is governed by several criteria. First, since the bulk of first antibodies have been reported to be raised in guinea pigs, rabbits, sheep or goats, other species are required for precipitating sera. Second, precipitating antisera are usually of low titre, operating at initial dilution of 1:30, so requiring the provision of large volumes of antisera and consequently the use of large animals. These criteria are usually satisfied by the use of donkeys, sheep or goat. Large and mature animals are therefore recommended (Seibelsch and Andrian 1969; Erlanger

et al 1999). In our study we used two sheep, kept at MISAR's animal house. The animals were fed on mixed green grass and grains (gram seeds). Two to three litre bloods were taken within seven days after each boost. Blood was collected from jugular vein by venepuncture after each booster injection. The serum was separated and preserved with 1% sodium azide. Unfortunately one sheep died in the beginning because of disease and study could only be continued in one sheep.

Testing of second antibody in sheep and it's optimisation (dilution curve experiments): As discussed in production of primary antibody optimum dilution of the second antibody is also required to set up good quality assays if we are employing second antibody precipitation technique to separate free and bound fractions. The design of dilution curve we followed was similar to that used for dilution experiment for primary antibody. An optimised quantity of non-immune serum from the appropriate first antibody species is normally added to increase the precipitate and simplify the separation.

Reagents prepared: (i) *Assay buffer* (glycerophosphate buffer): pH 7.3, 0.03M, 1%BSA, sodium azide/0.02% (v/v) ¹²⁵I-T₄/AMG solution (>35000 cpm/50 µl). Diluted the concentrated solution of the tracer using assay buffer to get approximately 35000 counts per 50 µl volume. Added 2 mg per ml ANS into this solution. (ii) 1% Polyethylene Glycol, MW=4000 (PEG). Dissolved 4 grams of PEG in 100 µl assay buffer. (iv) T₄ solution of concentration, 20 ng/ml.

Materials arranged: (1) Pure sheep anti-rabbit serum (SARS), (2) Non-immune rabbit serum (3) Polystyrene tubes (4) Assay buffer (5) Gamma counter.

Procedure: Following dilutions of sheep anti-rabbit sera prepared in assay buffer: 1:2 1:4, 1:8, 1:16, 1:32, 1:64, 1:128, 1:256, 1:512. As previously described two sets of tubes were prepared for setting up optimisation dilution experiment. Added 50 µl of ¹²⁵I-T₄ (>35000 cpm) into each tube. Added 50 µl of assay buffer into each tube of set-1 and then 50 µl of T₄ solution of conc. 20 ng/ml into each tube of set-2. All of these were vortexed. 50 µl of non-immune rabbit serum was also added into each tube of both the sets. The mixtures were kept at room temperature overnight (18-24h) followed by an additional incubation of 1 h at 4 °C. 0.5 ml of 4% PEG was finally added to precipitate the complex. The mixtures were finally centrifuged for 30 minutes at 10000 rpm, decanted and counted.

Preparation of working T₄ standards: Standard material was prepared in a way similar to that described for T₄. The working standards were prepared in hormone free serum. A stock standard of concentration >40.3 nmol T₄/l was prepared from a solution of T₄ of concentration, 1287 nmol/l (1 mg/ml)

using assay buffer as-diluent (Chapman *et al.* 1983; Jones and Perry 1985; Malan *et al.* 1983). This stock was further diluted to concentrations 1/1.5, 1/6.8, 80/4, 40/2, 20/1, and 10 nmol T_c/l respectively using T_c-free serum. T_c-free serum was also used as zero or control standard.

Quality control sera: Quality control sera of low, medium and high concentration were purchased from Amersham, U.K. The range of expected values as supplied by the manufacturer is given in Table 2. (Chapman *et al.* 1983; Jones and Perry 1985; Malan *et al.* 1983).

T_c Radioimmunoassay: using PEG assisted 2nd antibody precipitator; Assay design and protocols. Reagents and Other Items Required: (1) Standard Solutions. Prepared above (2) Anti-T_c-antibody at dilution 1:30000 (3) ¹²⁵I-T_c solution; conc. ~14000cpm/25 μ l-prepared above (4) Anti-rabbit antibody or second antibody—The pure anti-serum was diluted using assay buffer to a dilution of 1:60 (5) 4% PEG in assay buffer (6) Polystyrene assay tubes (1cm x 8cm). (7) Well type gamma-counter.

Assay design. Labelled the assay tubes in following sequence.

Tube No. Specification

- 1-2 Total activity tubes (T)
- 3-4 Non specific binding tubes (NSB)
- 5-6 0.0 nmol/l Calibrator (C1)
- 7-8 10 nmol/l Calibrator (C2)
- 9-10 20 nmol/l Calibrator (C3)
- 11-12 40.2 nmol/l Calibrator (C4)
- 13-14 80.4 nmol/l Calibrator (C5)
- 15-16 160.8 nmol/l Calibrator (C6)
- 17-18 321.5 nmol/l Calibrator (C7)
- 19-20 QC-1 (Quality control serum of low concentration)
- 21-22 QC-2 (Quality control serum of medium concentration)
- 23-24 QC-3 (Quality control serum of high concentration)
- 25-26 Sample-1, placed a total of 30 samples in duplicate until tubes 83-94
- 85-86 QC-1
- 87-88 QC-2
- 89-90 QC-3
- 91-92 Sample-2, placed a total of 30 samples in duplicates until tubes 119-140
- 93-94
- 115-116
- 117-118
- 119-120
- 121-122 QC-1

- 143-144 QC-2
- 145-146 QC-3

Assay Procedure: Added 50 μ l of standard, unknown and control sera into appropriate tubes. 50 μ l of rabbit anti-T_c antibody solution was added into all tubes-except totals and NSB tubes where 50 μ l assay buffer was added. 50 μ l of ¹²⁵I-T_c (30000cpm/ANB=1mg/ml), 50 μ l of normal rabbit serum (NRS) and 50 μ l of second antibody were then added into each tube. Incubated the mixture at room temperature overnight (18-21h). 0.5 ml of 4% PEG at 4 °C was added at the end of incubation. Vortexed and allowed the mixture to stand for 15 minutes at 4 °C and then centrifuged the contents. Counted the precipitates (bound fraction) for 90 seconds.

Data analysis: The data analysis philosophies of Dudley (1981), Radford *et al.* (1978) and Elkins (1978-81) were followed. The programs developed by WHO, were used for data analysis of T_c assay.

Results and discussion

Production of anti-T_c: Cross-reactivity. The T_c levels measured before and after dialysis are given in Table 1. These show that about 76% of the added T_c was coupled to the albumin and retained in the reaction mixture. The T_c-albumin conjugate concentration in the final mixture was approximately 1.56 mg/ml. The quantity produced was therefore sufficient for a number of immunizations. The low recovery compared to the expected values may be due to experimental errors and impurities in the commercial preparations.

Antibody titer: The antibody dilution curves of some of the rabbits obtained from sera of different rabbits at 3 weeks intervals are shown in Figs 4-6. The antibody titres observed at 50% tracer binding are given in Table 3. Studies in all rabbits could not be continued because of poor response and mortalities due to disease and unfavorable climatic

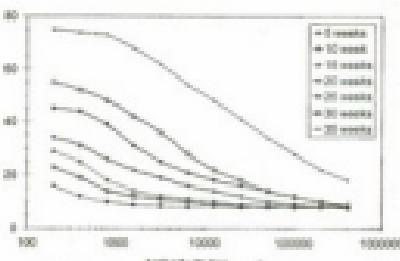


Fig. 4. T_c assay (Rabbit). Antibody dilution curves. Blood samples collected at 3 weeks intervals.

Table 2
Typical assay data obtained on a T₁-RIA, applying second antibody precipitation

Tube No.	Specification	CPMI	CPMII	Mean	%CV	%SNT
12	T	16551	16463	16506	1.1	—
34	NSB	889	679	622	5.2	1.8
56	C1 (100%)	21941	20903	20761	1.8	60.3
78	C2 (10%)	19340	18129	18680	0.5	59.80
9-10	C3 (20.1%)	19340	18984	19143	1.8	44.2
11-12	C4 (40.2%)	11449	10328	11486	0.9	33.9
13-14	C5 (80.4%)	7889	6010	7054	1.0	23.2
15-16	C6 (100.0%)	4580	4300	4430	2.2	12.9
17-18	C7 (211.5%)	2650	2380	2415	1.9	7.6
19-20	QC1	16690	16888	16794	1.0	40.1
21-22	QC2	8284	8253	8263	1.1	24.8
23-24	QC3	4723	4690	4702	0.6	11.7
25-26	U-1	2361	2095	2076	6.8	6.68
27-28	U-2	3072	3083	3128	1.5	9.12
29-30	U-3	4011	4145	4028	8.7	11.5
31-32	U-4	9803	9413	9604	1.2	21.7
33-34	U-5	5663	5795	5699	0.7	15.95
35-36	U-6	14876	14779	14628	2.4	42.65
37-38	U-7	6217	7031	6876	2.6	26.69
39-40	U-8	7077	7413	7445	0.6	21.70
41-42	U-9	9711	9824	9767	0.8	28.5
43-44	U-10	3807	3848	3853	4.2	11.47
45-46	U-11	10752	14793	14773	0.2	43.07
47-48	U-12	7229	7615	7420	3.7	23.64
49-50	U-13	6400	6290	6349	1.8	18.99
51-52	U-14	19338	17993	18174	1.4	52.99
53-54	U-15	7029	7073	7041	0.5	20.56
55-56	QC1	14543	14958	14850	1.0	40.3
57-58	QC2	8291	8473	8382	1.5	24.4
59-60	QC3	4756	4696	4710	0.5	11.7
61-62	U-16	9101	9364	9338	0.4	22.3
63-64	U-17	6813	7041	6787	3.3	18.79
65-66	U-18	4751	5043	4897	4.2	10.28
67-68	U-19	3254	3232	3242	0.5	9.46
69-70	U-20	3794	4016	3878	5	10.31
71-72	U-21	7007	7439	7209	2.4	22.07
73-74	U-22	16014	16607	16190	1.1	47.21
75-76	U-23	8826	8704	8815	2.3	19.87
77-78	U-24	8359	8873	8615	4.2	25.12
79-80	U-25	8289	9287	9409	1.8	25.44
81-82	U-26	8234	8520	8396	1.6	24.57
83-84	U-27	9159	9245	9200	0.6	26.80
85-86	U-28	7648	7962	7760	3.2	22.71
87-88	U-29	7086	7025	7054	2.4	22.06
89-90	U-30	6971	7012	7021	3.6	20.85
91-92	QC1	14530	14642	14586	0.5	42.9
93-94	QC2	8159	8070	8093	0.9	23.6
95-96	QC3	4760	4680	4693	1.3	12.8

The results of antibody titre varied in rabbits. Rabbit-1, 1kg, male, actively responded to the immunogen (Fig 4). The response was very steep after fourth boost. A maximum titre of 1: 9700 was observed after 15 weeks. Rabbit-2 responded excellently but could not survive after 2nd booster injection although the titre at that time was 1:8000 (this seemed to be showing the most active response). Rabbit-3 responded posi-

tively but could not live long. Rabbit-10 (1.5Kg, female) showed a very good response (titre, 1:2000). Rabbit-12 (1.0Kg female) also responded very well (Fig 5) and the observed titre after 7th boost was 1:4000. Studies on other rabbits could not be continued because some of them died and others did not respond to the immunogen very well. Response in some of the rabbits was very low e.g., rabbit-4 (Fig 6). All this suggests

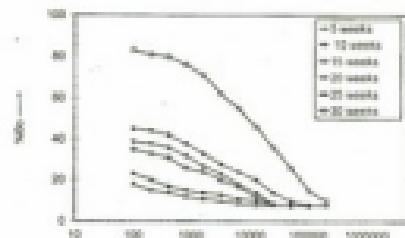


Fig 5. Titre assay: Rabbit-12. Antibody dilution curves. Blood samples collected at 3 weeks intervals.



Fig 6. Titre assay: Rabbit-4. Antibody dilution curves. Blood samples collected at 3 weeks intervals.

Table 3
Titre Antibody titres in rabbits determined 23 weeks after the primary injection

Rabbit identification	Weeks after primary injection	Titre*	Remarks
Rabbit-1, male, 1.5 Kg body weight	15	1:9700	Continued to live for five years when it was finally killed. It gave high quantities of antibodies, which could be useful for future work.
Rabbit-2, male, 1.2 Kg body weight	10	1:8000	Gave good response but died after 10 weeks.
Rabbit-3, female, 0.8Kg body weight	15	1:2500	This rabbit showed a good response to the immunogen and developed significant quantities of antibodies but died because of infection. Could not survive.
Rabbit-4 male, 1.0 Kg body weight	12	1:300	Could not survive long.
Rabbit-5 female, 1.0 Kg body weight	10	low	Could not survive
Rabbit-6, female, 1.6 Kg body weight	10	low	Could not survive
Rabbit-7, male, 1.6 Kg body weight	5	low	Responded very low
Rabbit-8, female, 1.6 Kg body weight	15	low	Response very low
Rabbit-9, female 800 g body weight	10	low	Could not survive
Rabbit-10, female 910 g body weight	15	1:2000	Good response
Rabbit-11, female 800 g body weight	5	low	Response poor
Rabbit-12, female 800 g body weight	10	1:200	Good response

*Titre is defined as the dilution at 50% binding of tracer with the antibody.

that at least four rabbits had responded very well to the immunization. Serum from all the test bloods was collected and mixed with sodium azide to prevent bacterial growth and stored at -20 °C. Antibody of Rabbit-1 was first used for future experimental work. The very high titer of the antibody suggest that the home made antibody could be sufficient for millions of tests.

Optimization of antibody concentration. The dilution curve obtained by plotting bound counts in two sets of tubes against antibody dilution (rabbit-1) is shown in Fig 7. Bo-Curve shows the binding of the tracer to the antibody in the absence of any unlabelled T_1 . It can be appreciated that about 50% of the tracer is tagged to the antibody at the point of maximum binding. The Bi-Curve i.e., the curve obtained in the presence of unlabelled T_1 , splits well from the Bo-curve in the middle, which suggests that the avidity of the antibody is high and a sensitive assay can be developed from this antibody.

An error curve can be constructed from these two curves by plotting the reciprocal of the difference of counts between two curves at various dilutions against the dilution of the primary antibody. (The ratio Bi/(Bo-NBi) can also be plotted against the antibody dilution, as we have done in T_1 assay). The curve thus obtained is shown in Fig 8, which depicts that the antibody will work very well from dilutions 1:1000 to 1:10000. The minimum error point is around 1:8000, which was selected as the optimum antibody dilution. This confirms that there were significant quantities of antibody in our rabbit serum and that many optimum assays could be developed using this cheap locally developed antibody. According to our initial estimates the quantity of serum we have collected could be sufficient for millions of tubes. (i.e., 1 CC serum will be sufficient for about 100000 tubes).

In conclusion the local techniques were successful in producing antibody of high titer which is very sensitive to

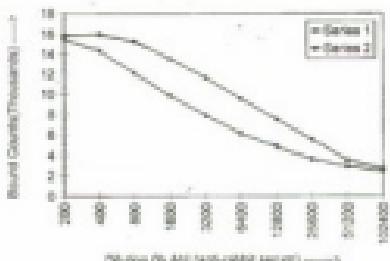


Fig. 7. T_1 assay: Optimization of primary T_1 antibody (Bo and Bi) against antibody dilution.

low concentrations of T_1 . This antibody could develop many low cost assays.

Preparation of iodinated tracer for T_1 and T_2 RIA. The radiochromatogram obtained after purification through sephadex is shown in Fig 9. The analysis of counting data (Table I) showed that 31% of the total activity appears as $^{131}\text{I}-T_1$, 30% as $^{131}\text{I}-T_2$ and 19% as free iodide. 4% of total activity is lost as contaminations. The specific activities of these tracers calculated from counting data of iodination experiments were: $T_1 = 300-400 \mu\text{Ci/mg}$, $T_2 = 50-60 \mu\text{Ci/mg}$.

Production of second antibody assay. The dilution observed in sheep after booster injections are shown in Table I. The maximum dilution, which we obtained, was 1:60. The sheep did not respond more after 4th booster.

Optimization of second antibody: The dilution curves obtained after analysis of counting data of one of our experiments on optimization of second antibody is shown in Fig 10.

The system seems to be very sensitive to T_2 and there is a very good splitting of the two curves near the maximum binding. If we correlate this splitting (i.e., the reciprocal of the

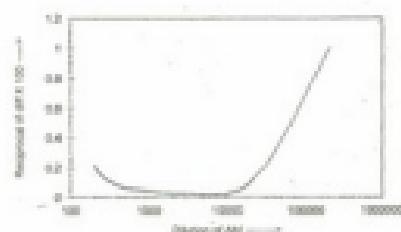


Fig. 8. T_1 assay: Testing of anti-rabbit serum for optimum antibody dilution by 12% PEG precipitation.

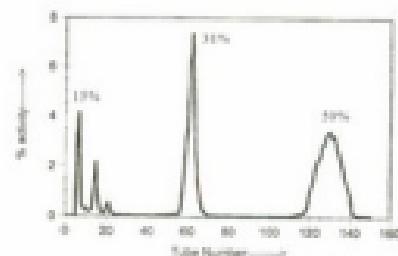


Fig. 9. Radiochromatogram of iodination products obtained after purification through sephadex column. Radiochromatogram

difference in the count rate of the two curves/ λ 10% with the antibody dilution, we get another curve shown in Fig. 11. Here the lowest point indicated the dilution where the sensitivity of the system will be highest. This may also be called the minimum error point. As can be read from the curve the dilution of second antibody at this point is 1:60. This is therefore the working dilution of the second antibody.

T₁-Radioimmunoassay using PEG assisted 2nd antibody precipitation. Typical assay data obtained using above protocol is given in Table 2. The assay statistics and curve parameters obtained using WHO data reduction programs are given in Table 3.

The standard curve plotted using the counting data obtained in above protocol is shown in Fig. 12. The curve parameters show that the standard data is perfectly fit in logit-log system of coordinates. The non-specific binding is about 2%. This is consistent with findings of others (Edwards 1983; Edwards et al 1985). The use of second antibody therefore improved the separation in our T₁ and T₂ assays. The response error relationship and precision profiles are shown in Fig 13 and 14 respectively. The curves show more than 7% error in the concentrations below 10

nmol/l. The system is also performing well at higher concentrations and has therefore a very wide working range.

The observed values of quality control sera are given in Table 6. The values are within the expected range. Mean within assay drift is acceptable. The correlation of the values using this in house technique with those obtained using commercial kits is shown in Fig 15. The assay correlates (Correlation Coefficient=0.99) well with the commercial techniques.

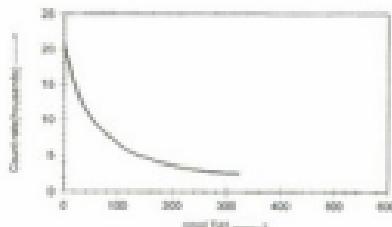


Fig. 12. T₁-RIA standard curve (double antibody precipitation).

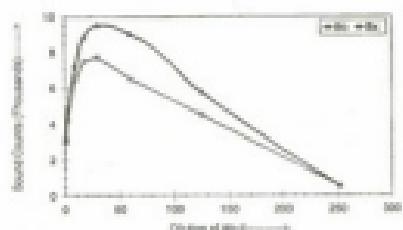


Fig. 13. T₁ assay: T₂ optimization of second antibody. Total counts = 28815 cpm; NSB = 228

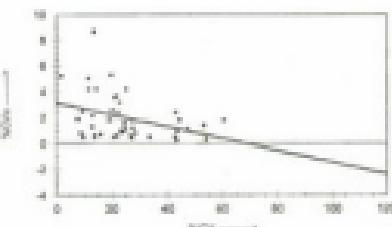


Fig. 14. T₁ RIA (double antibody method): Response error relationship.

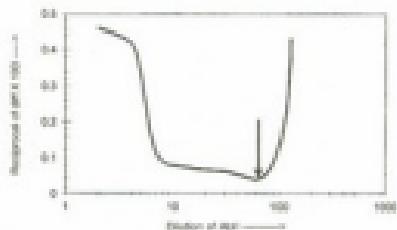


Fig. 15. T₁ assay: Optimization of second antibody. The -one curve. Total counts = 28815 cpm; NSB = 228

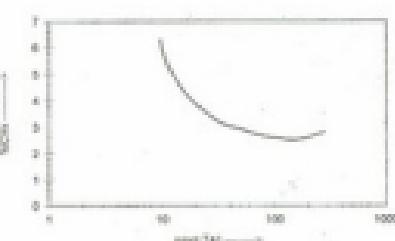


Fig. 16. T₁ RIA (double antibody): Inhibition profile.

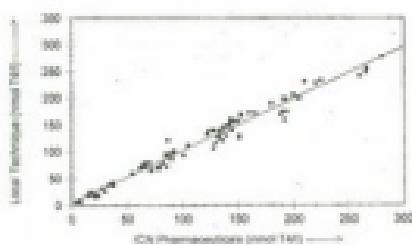


Fig. 12. Correlation of local T_1 values (double antibody assay) with commercial kit. Corr. coeff. = 0.99, Regression $a = 4.97$, $b = 0.97$.

Table 4

Anti-rabbit antibody titres in sheep determined in first four months after primary injection

Months	First	Second	Third	Fourth
Antibody titre	1:20	1:30	1:60	1:60

Table 5

Double antibody T_1 -assay: Overall assay statistics

Assay counting statistics	Minimum%CV=0.2
	Minimum%CV=0.7
	Mean%CV=1.99
	Median%CV=1.3
RER parameters	$A=0.15$
	$B=0.066$
Curve parameters	Logit-Log corr. coefficient = -0.9947
	Logit-log slope = -1.151
	Logit-log intercept = -4.933
	ED90=50.82
	ED99=4.109

Table 6

Double antibody T_1 -RIA: Expected and observed values of quality control sera

Specification	Expected Values (nmol T_1 /l)		Observed Values (nmol T_1 /l)	
	Mean	Range	Mean	Range
QC-1 (Low control)	20.7	16.5-24.9	21.10±0.7	
QC-2 (Medium)	68.3	54.6-82	73.3±1.5	
QC-3 (High)	177	143-213	162.5±3.2	

To summarize, the locally developed RIA assisted double antibody assay system was found to be working very well at all clinical regions (hypo, hyper, normal). The assay precision was excellent and the system seemed to fit well on all Standards of clinical assessment.

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